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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=296 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.005 \AA$
$R$ factor $=0.071$
$w R$ factor $=0.198$
Data-to-parameter ratio $=13.5$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
## 3-[3-(Trifluoromethyl)anilino]isobenzo-furan-1(3H)-one

Crystals of the title compound, $\mathrm{C}_{15} \mathrm{H}_{10} \mathrm{~F}_{3} \mathrm{NO}_{3}$, are stabilized by $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ intermolecular hydrogen bonds and by weak $\mathrm{C}-\mathrm{H} \cdots \pi$ interactions. In the structure, paired $\mathrm{C}-$ $\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds link the molecules into $R_{2}^{2}(10)$ dimers. The hydrogen-bonded $R_{2}^{2}(10)$ dimers are linked by intermolecular $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds and generate $R_{6}^{4}(22)$ rings. The phthalide section of the molecule is planar and the dihedral angle between the phthalide group and the benzene ring is $56.31(17)^{\circ}$.

## Comment

The present work is part of a structural study of compounds of 3-substituted phthalides and we report here the structure of 3-[3-(trifluoromethyl)anilino]isobenzofuran-1(3H)-one, (I) (Fig. 1).

(I)

The phthalide group (atoms $\mathrm{C} 1-\mathrm{C} 8 / \mathrm{O} 2$ ) is essentially planar, the largest deviation from the mean plane being 0.042 (3) $\AA$ for atom C8. The dihedral angle between the mean planes of the phthalide group and the trifluoromethylphenyl ring is $56.31(17)^{\circ}$. This angle is $54.55(10)^{\circ}$ in 3-[2-(trifluoromethyl)anilino]phthalide (Odabaşoğlu \& Büyükgüngör, 2006a), $78.43(15)^{\circ}$ in 3-anilinophthalide (Odabaşoğlu \&


Figure 1
A view of (I), showing the atomic numbering scheme, with displacement ellipsoids drawn at the $30 \%$ probability level.

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3-Substituted phthalides, Part XIX


A packing diagram for (I). Dashed lines indicate hydrogen bonds.


Figure 3
Part of the crystal structure of (I), with hydrogen bonds drawn as dashed lines, showing the formation of a hydrogen-bonded dimer. For the sake of clarity, H atoms bonded to C atoms have been omitted. [Symmetry code: (i) $1-x, 1-y, 1-z]$.

Büyükgüngör, $2006 b$ ), and $54.55(10)^{\circ}$ in 3-(4-acetylanilino)phthalide (Odabaşoğlu \& Büyükgüngör, 2006c).

The crystal packing is stabilized by $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{N}-$ $\mathrm{H} \cdots \mathrm{O}$ intermolecular hydrogen bonds, which generate $R_{2}^{2}(10)$ (Fig. 2) and $R_{6}^{4}(22)$ rings (Fig. 3) (Etter, 1990), and also by C$\mathrm{H} \cdots \pi$ interactions (Fig. 4, Table 2).

Figure 4


A packing diagram for (I), showing the $R_{6}^{4}(22)$, hydrogen bonds and $\mathrm{C}-$ $\mathrm{H} \cdots \pi$ interactions represented as dashed lines. H atoms not involved in hydrogen bonds have been omitted for clarity. [Symmetry codes: (i) $1-x$, $1-y, 1-z$; (ii) $1-x, 1-y, z-\frac{1}{2}$; (iii) $\left.x, y-1, z+\frac{1}{2}\right]$.

## Experimental

The title compound, (I), was prepared as described by Odabaşoğlu \& Büyükgüngör ( $2006 d$ ), using phthalaldehydic acid and 3 -fluoromethylaniline as starting materials (yield $78 \%$; m.p. 417-418 K). Crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of a glacial acetic acid solution at room temperature.

## Crystal data

$\mathrm{C}_{15} \mathrm{H}_{10} \mathrm{~F}_{3} \mathrm{NO}_{2}$
$M_{r}=293.24$
Monoclinic, $P 2_{1} / c$
$a=13.4659$ (9) A
$b=8.1794$ (7) $\AA$
$c=12.2649$ (8) A
$\beta=97.400(5)^{\circ}$
$V=1339.64(17) \AA^{3}$

## Data collection

Stoe IPDS-2 diffractometer
$\omega$ scans
Absorption correction: integration

$$
(X-R E D 32 ; \text { Stoe \& Cie, 2002) }
$$

$T_{\text {min }}=0.942, T_{\text {max }}=0.986$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.071$
$w R\left(F^{2}\right)=0.198$
$S=1.07$
2629 reflections
195 parameters
H atoms treated by a mixture of independent and constrained refinement

$$
\begin{aligned}
& Z=4 \\
& D_{x}=1.454 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \mu=0.13 \mathrm{~mm}^{-1} \\
& T=296 \mathrm{~K} \\
& \text { Prism, colourless } \\
& 0.61 \times 0.42 \times 0.12 \mathrm{~mm}
\end{aligned}
$$

14507 measured reflections 2629 independent reflections 1933 reflections with $I>2 \sigma(I)$

$$
R_{\mathrm{int}}=0.104
$$

$$
\theta_{\max }=26.0^{\circ}
$$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0685 P)^{2}\right. \\
& +1.3237 P] \\
& \text { where } P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}{ }^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.001 \text { 。 } \\
& \Delta \rho_{\max }=0.40 \mathrm{e}_{\AA^{-3}} \\
& \Delta \rho_{\text {min }}=-0.29 \mathrm{e}^{-3} \\
& \text { Extinction correction: SHELXL97 } \\
& \text { Extinction coefficient: } 0.011 \text { (3) }
\end{aligned}
$$

## organic papers

Table 1
Selected geometric parameters $\left(\AA^{\circ},{ }^{\circ}\right)$.

| $\mathrm{C} 1-\mathrm{O} 1$ | $1.203(4)$ | $\mathrm{C} 2-\mathrm{C} 3$ | $1.372(5)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{C} 1-\mathrm{O} 2$ | $1.344(4)$ | $\mathrm{C} 9-\mathrm{N} 1$ | $1.393(4)$ |
|  |  |  |  |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{O} 2$ | $122.0(3)$ | $\mathrm{N} 1-\mathrm{C} 8-\mathrm{O} 2$ | $111.5(3)$ |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 2$ | $128.8(3)$ |  |  |

Table 2
Hydrogen-bond geometry ( $\AA{ }^{\circ}{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{~N} 1-\mathrm{H} 1 \cdots \mathrm{O}^{\mathrm{i}}$ | $0.83(4)$ | $2.14(4)$ | $2.962(4)$ | $176(4)$ |
| $\mathrm{C} 8-\mathrm{H} 8 \cdots 1^{\mathrm{ii}}$ | 0.98 | 2.52 | $3.264(4)$ | 133 |
| $\mathrm{C} 8-\mathrm{H} 8 \cdots \mathrm{Cg} 1^{\mathrm{iii}}$ | 0.98 | 3.31 | $3.979(4)$ | 128 |
| Symmetry codes: | (i) $\quad x,-y+\frac{3}{2}, z+\frac{1}{2} ;$ | (ii) | $-x+1,-y+1,-z+1 ;$ | (iii) |
| $-x+1, y+\frac{1}{2},-z+\frac{1}{2}$. | $C g 1$ is the centroid of the five-membered ring |  |  |  |

The NH group H atom (H1) was found in a difference Fourier map and refined freely. All other H atoms were refined using the ridingmodel approximation, with $\mathrm{C}-\mathrm{H}=0.93$ for aromatic H atoms and $\mathrm{C}-\mathrm{H}=0.98 \AA$ for methine H atoms $\left[U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})\right]$.

Data collection: $X$-AREA (Stoe \& Cie, 2002); cell refinement: $X$ AREA; data reduction: X-RED32 (Stoe \& Cie, 2002); program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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